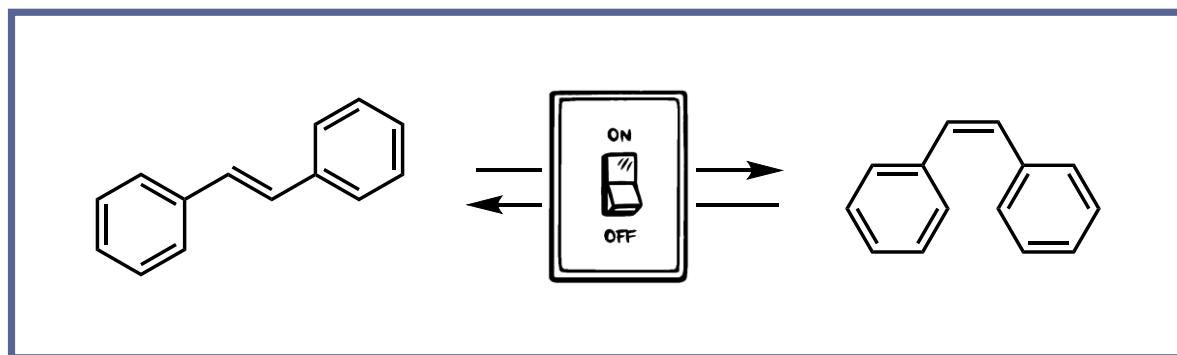


# Molecular Photoswitches

## Fundamentals and Applications



Angela Lin  
Knowles Lab Group Meeting  
Literature Talk  
January 13, 2023

# Outline

I. Introduction

II. Applications

- Photopharmacology
- Molecular Motors
- Solar Energy Storage

III. Conclusion & Outlook

# Outline

## I. Introduction

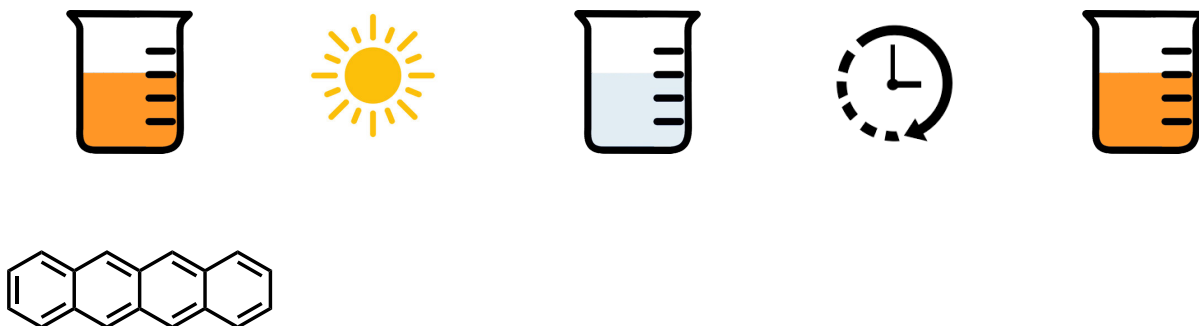
## II. Applications

- Photopharmacology
- Molecular Motors
- Solar Energy Storage

## III. Conclusion & Outlook

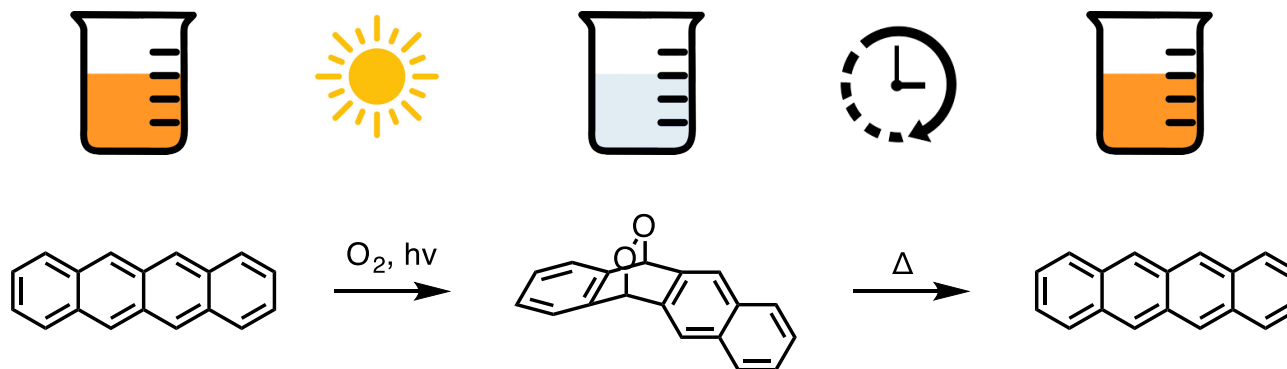
# Introduction

Photochromism is characterized by a light-induced reversible change in color, and was first described by Fritzsche in 1867.



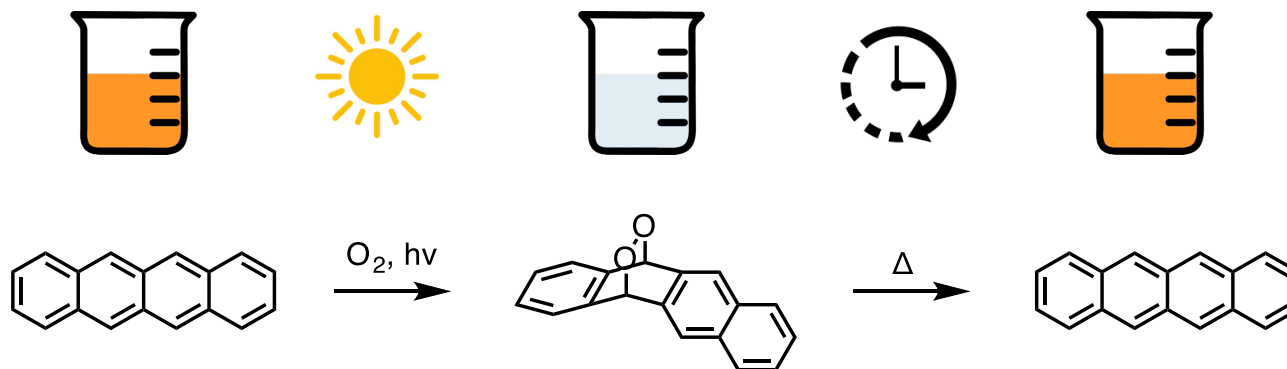
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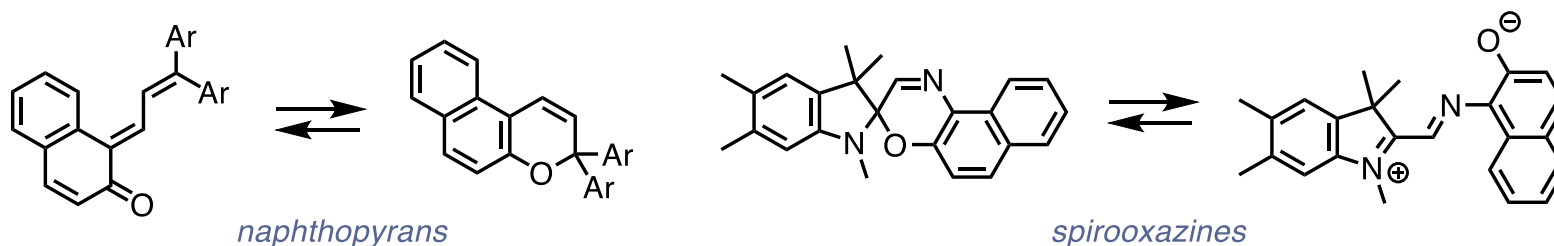
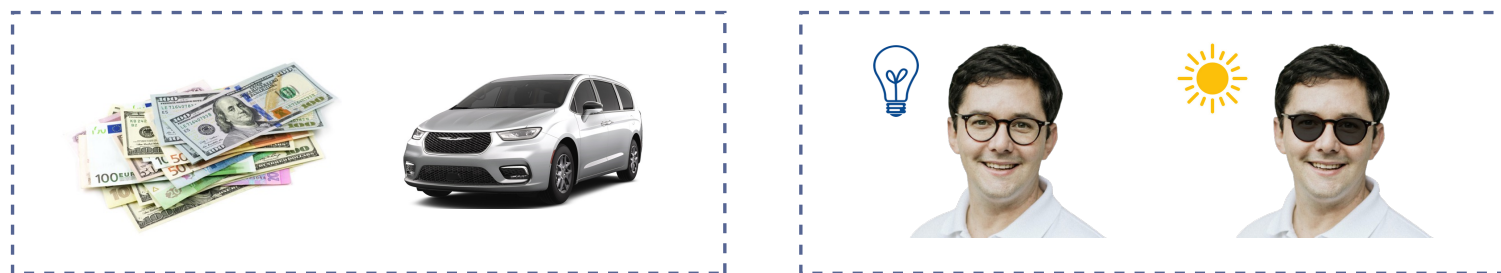


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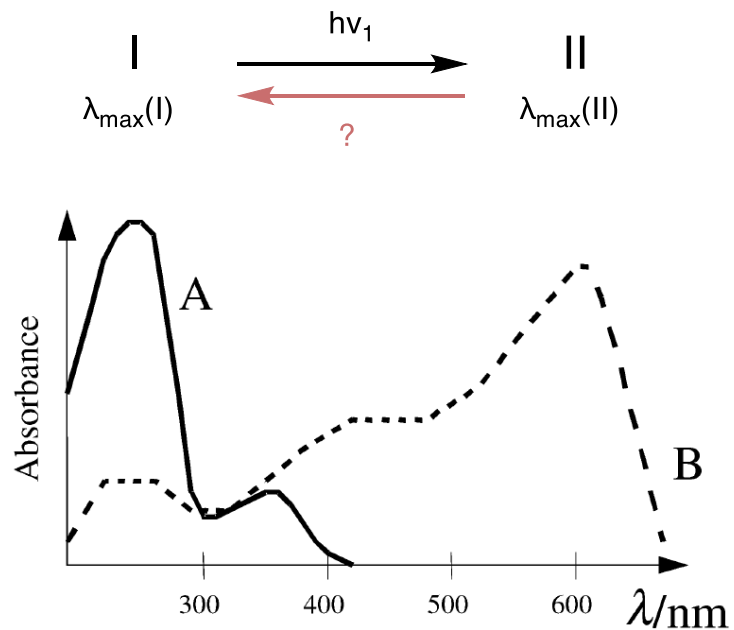
Since then, numerous photochromic compounds have been discovered and commercialized.



# Introduction

While photoswitches are indeed photochromic, the reversible changes they go through extend beyond just a simple color change.

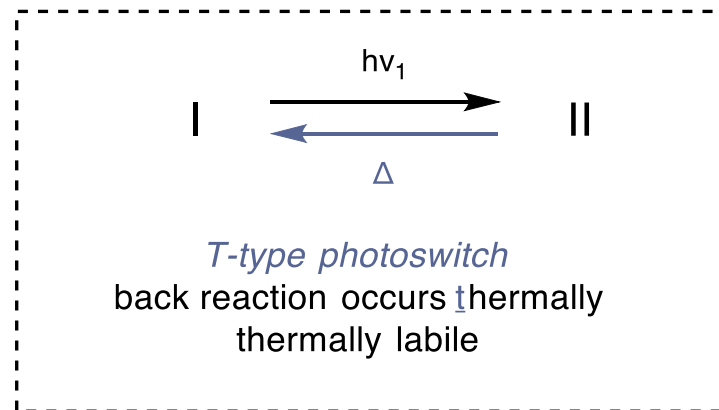
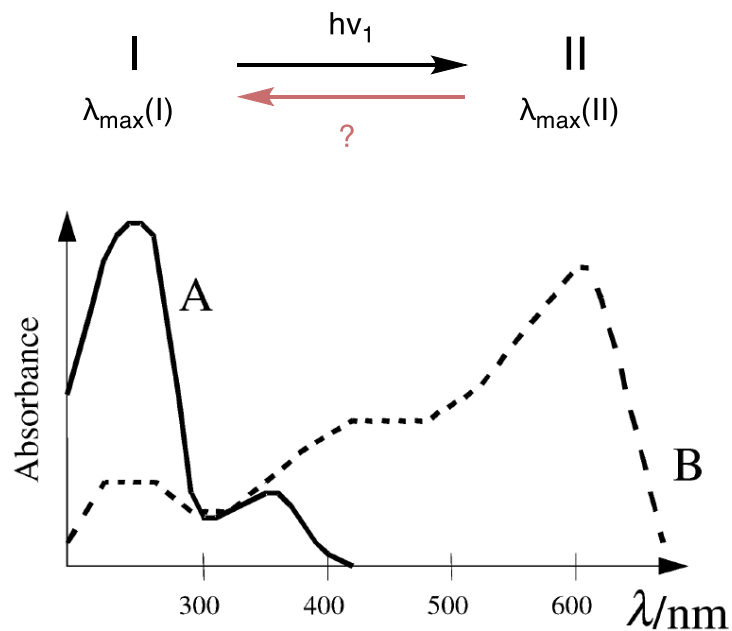
**Photoswitches** are small molecules that, upon exposure to light, undergo a reversible change in geometry, polarity, and/or charge distribution.



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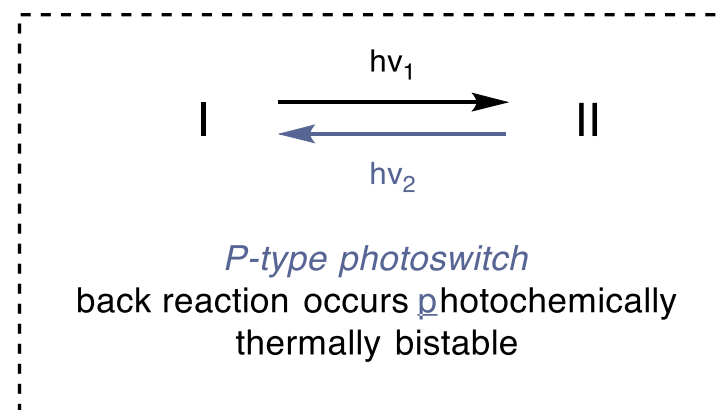
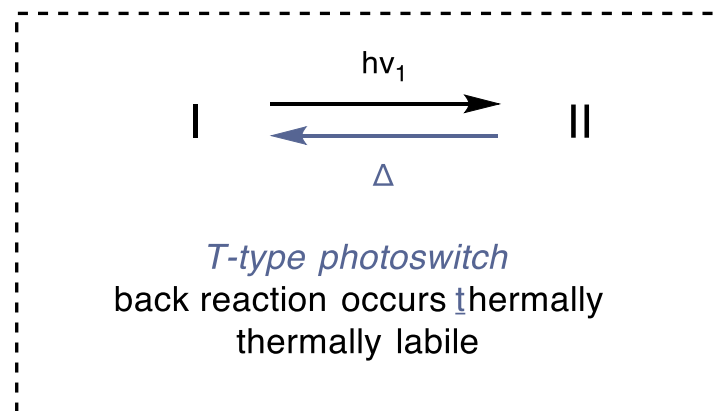
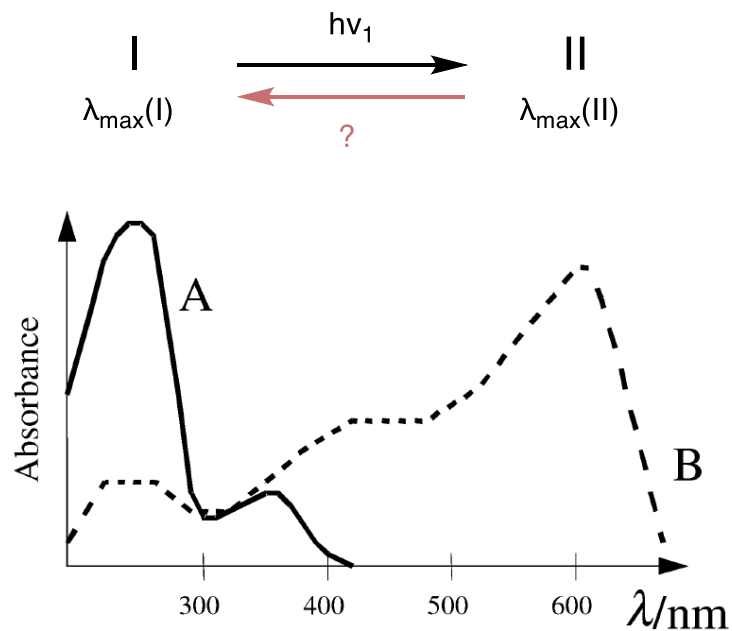




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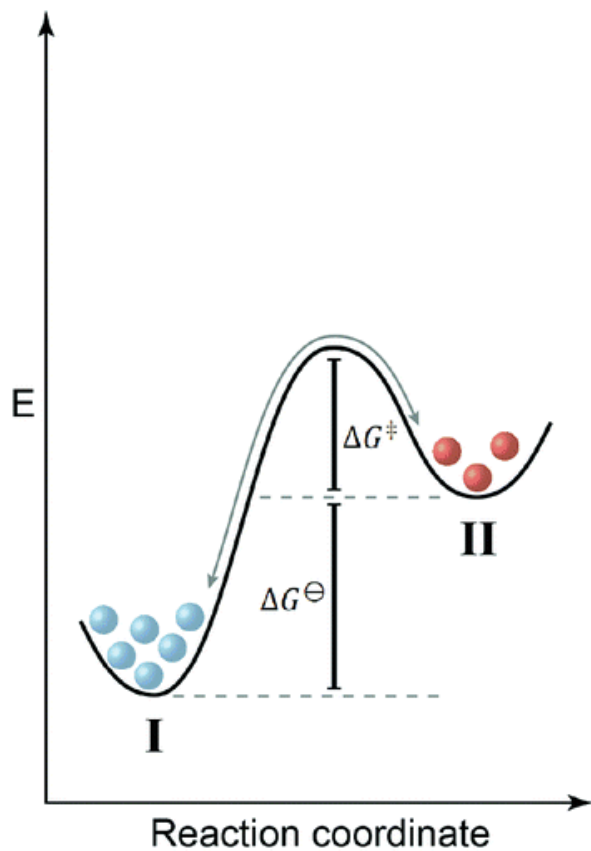
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**Photoswitches** are small molecules that, upon exposure to light, undergo a reversible change in geometry, polarity, and/or charge distribution.



# Thermodynamic vs. Photodynamic Equilibria

## Thermodynamic Equilibrium



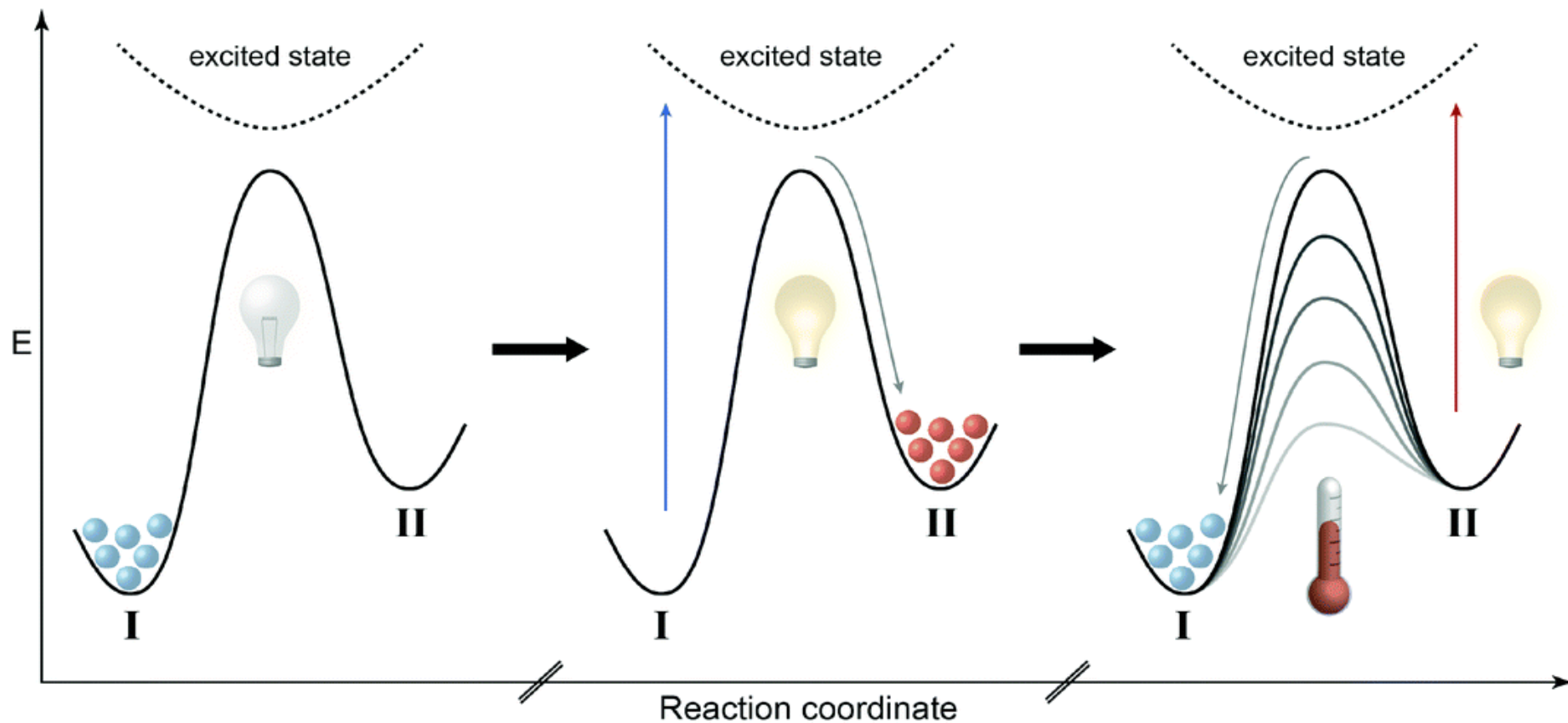
Thermodynamic equilibrium

$$K_{eq}^{\Delta} = \frac{II}{I} = e^{\frac{\Delta G^{\ominus}}{RT}}$$

Principle of microscopic reversibility

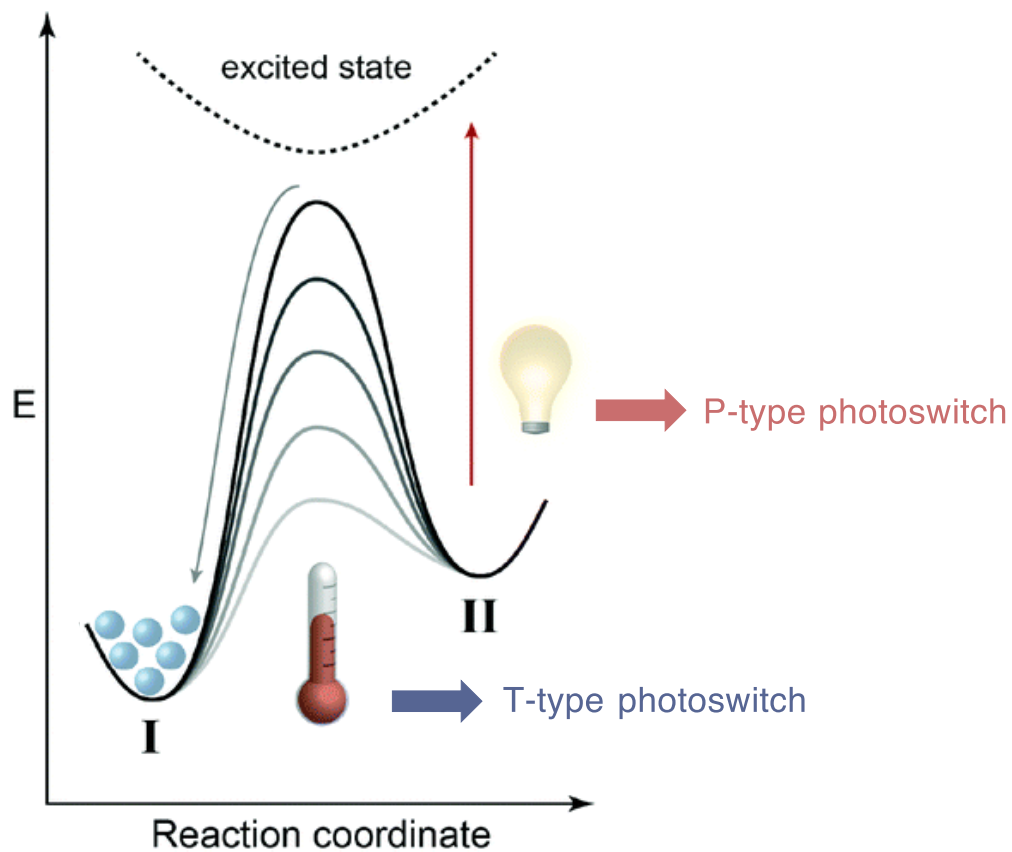
# Thermodynamic vs. Photodynamic Equilibria

## Photodynamic Equilibrium



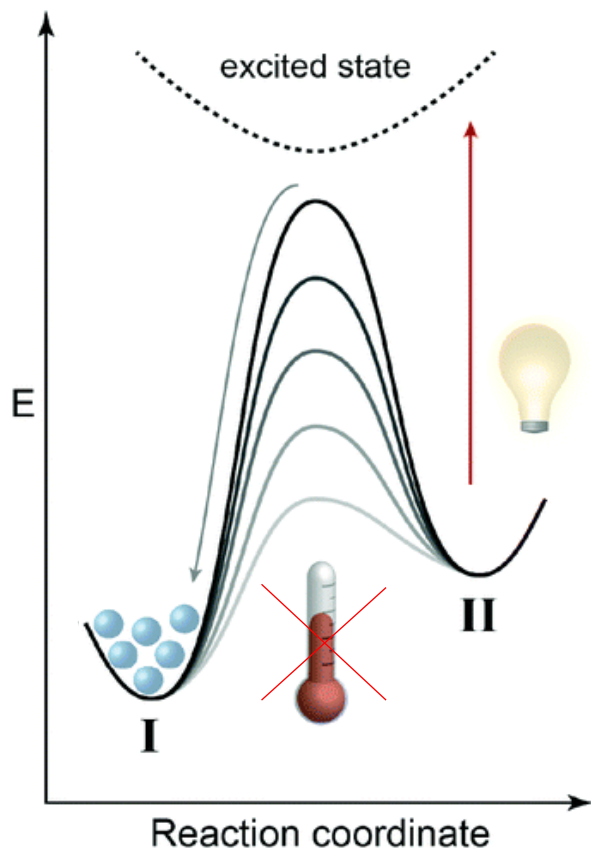
# Thermodynamic vs. Photodynamic Equilibria

## Photodynamic Equilibrium



# Thermodynamic vs. Photodynamic Equilibria

## Photodynamic Equilibrium



Photostationary state

$$K_{eq}^{\lambda} = \frac{II}{I} = \frac{\epsilon_I^{\lambda}(\Phi_{I \rightarrow II}^{\lambda})}{\epsilon_{II}^{\lambda}(\Phi_{II \rightarrow I}^{\lambda})}$$

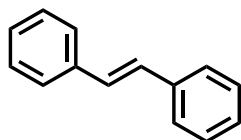
$\epsilon$  = extinction coefficient  
 $\Phi$  = quantum yield

Principle of microscopic reversibility does not apply

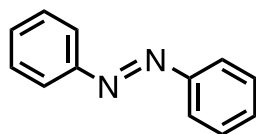
# Types of Transformations

Common photoswitches generally operate through 2 types of transformations:

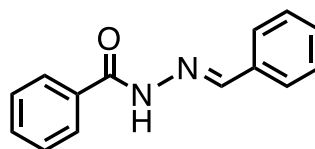
## Photoisomerization



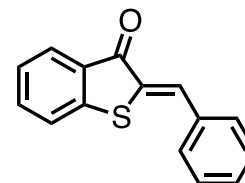
*stilbenes*



*azobenzenes*



*acylhydrazones*

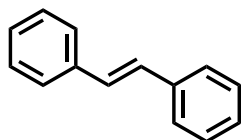


*hemithioindigo*

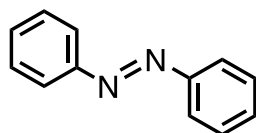
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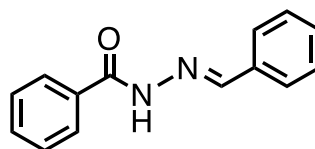
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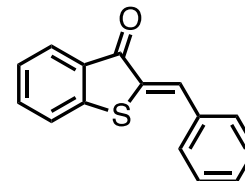
*stilbenes*



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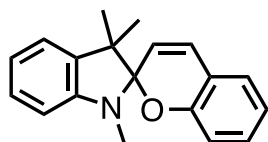


*acylhydrazones*

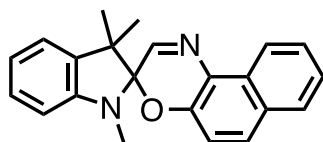


*hemithioindigo*

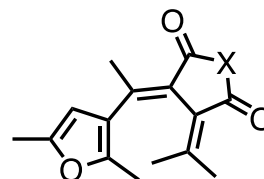
## Photocyclic ring-opening/closing



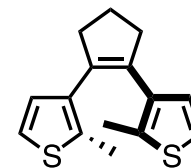
*spiropyrans*



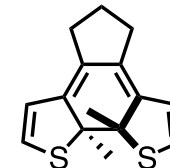
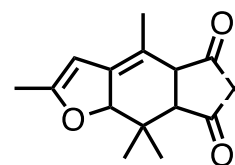
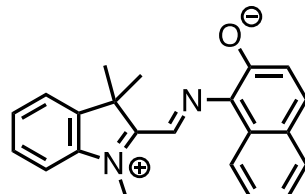
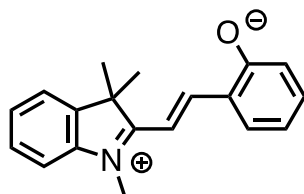
*spirooxazines*



*fulgides/fulgimides*



*diarylethenes*



\* Other (less common) processes include photo-induced proton transfer and dimerization.

# Design Considerations

- High quantum yields
- High extinction coefficients at the desired wavelength(s)
- High photostationary state
- Control over switching speed
- Tunable thermal stability of the isomers (half-life)
- Resistance against photodegradation (fatigue)



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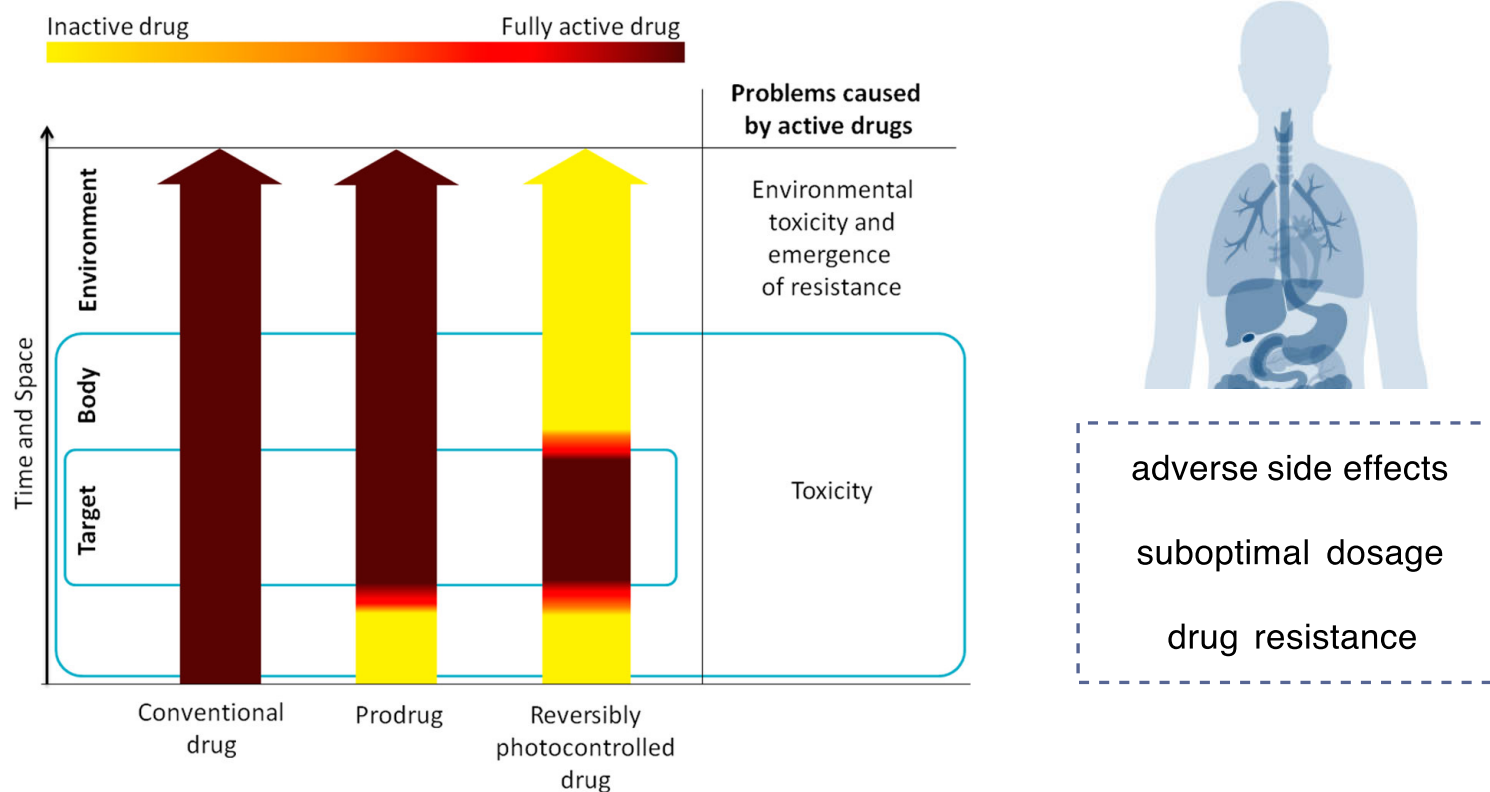
- Photopharmacology
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III. Conclusion & Outlook

# Photopharmacology

Photopharmacology involves the introduction of a photoswitch into the molecular structure of a bioactive compound.

Key issue in pharmacotherapy: lack of spatial and temporal control over drug activity



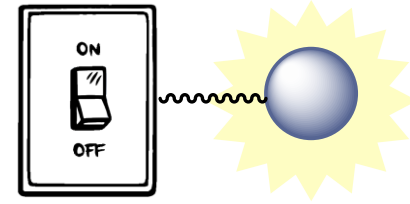
# Photopharmacology

## Advantages of light as an external stimulus

non-invasive

can be regulated as desired (wavelength, intensity)

can be delivered with high spatial and temporal precision

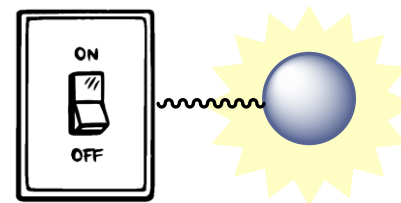


*reversible control*

# Photopharmacology

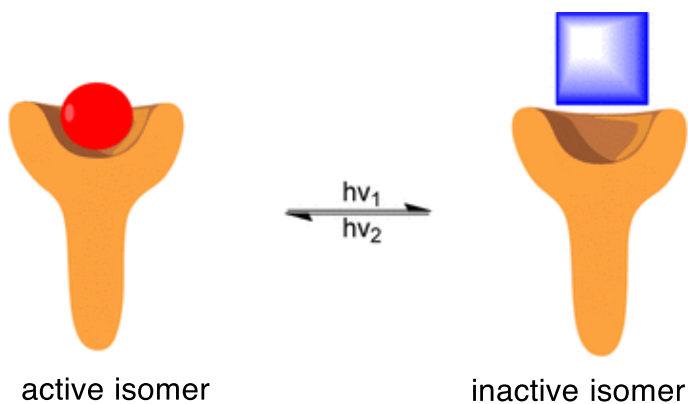
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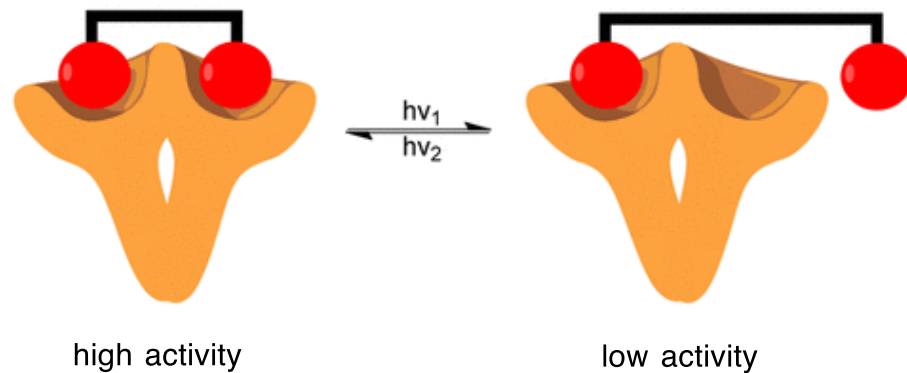


*reversible control*

## Strategy 1 (Direct)

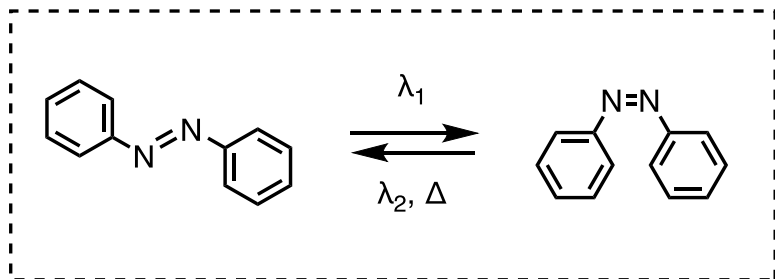


## Strategy 2 (Indirect)



# Popular Motifs

## Azobenzenes (T-type)



photoisomerization

large changes in geometry and polarity

$\lambda_{\max}(\text{E}) = 320 \text{ nm } (\pi \rightarrow \pi^*), 450 \text{ nm } (n \rightarrow \pi^*)$

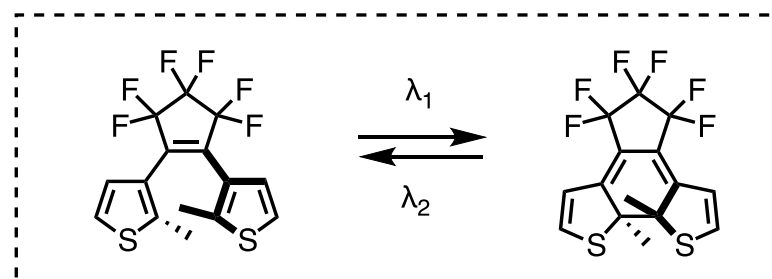
$\lambda_{\max}(\text{Z}) = 270 \text{ nm } (\pi \rightarrow \pi^*), 450 \text{ nm } (n \rightarrow \pi^*)$

$t_{1/2}(\text{Z}) = 2 \text{ days}$

distance change = 3 Å

dipole moment change = 3 D

## Diarylethenes (P-type)



photoinduced cyclization/ring opening

large changes in flexibility and electronics

$\lambda_{\max}(\text{open}) = 303 \text{ nm}$

$\lambda_{\max}(\text{closed}) = 505 \text{ nm}$

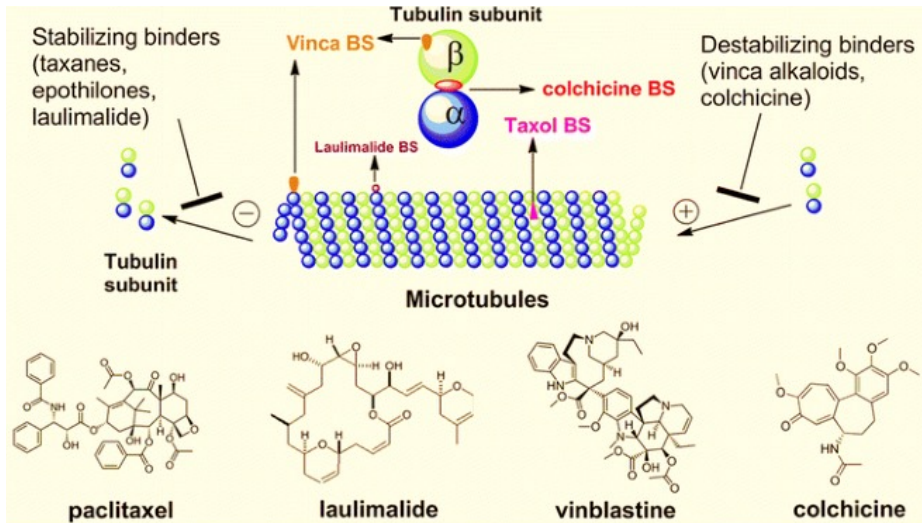
$t_{1/2}(\text{closed}) = \sim 4\text{-}5 \times 10^5 \text{ years}$

distance change = 1 Å

dipole moment change = minimal

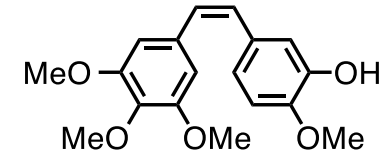
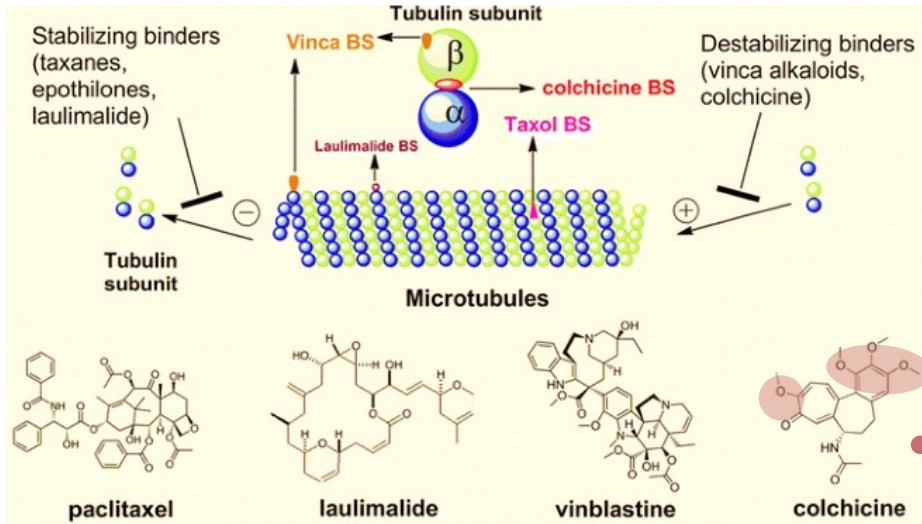
# Photostatins (PSTs)

Microtubule inhibitors are a class of chemotherapeutics known for antimetabolic and pro-apoptotic effects.



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Microtubule inhibitors are a class of chemotherapeutics known for antimetabolic and pro-apoptotic effects.



Combretastatin A-4 (CA4)

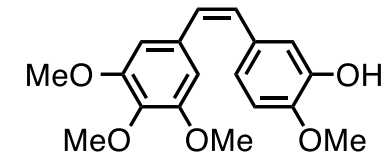
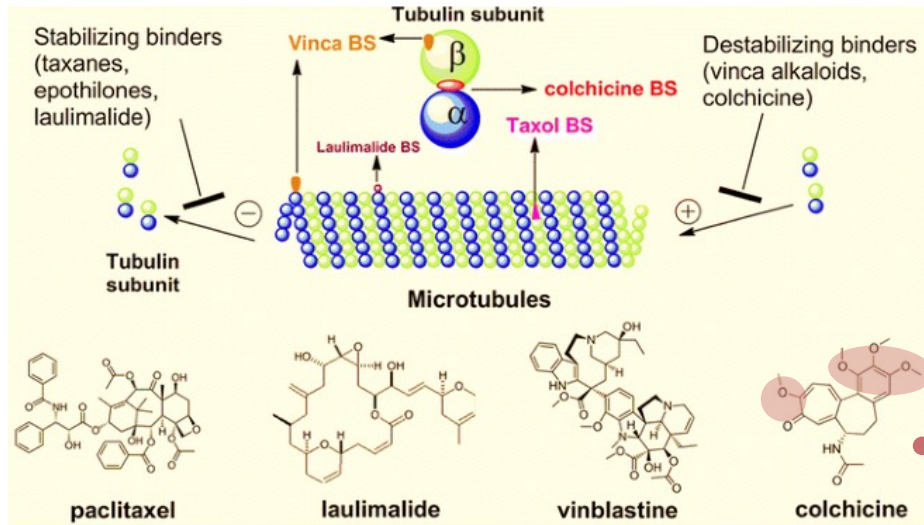
*multiple analogues have progressed to clinical trials*

*trans isomer is much less potent*



# Photostatins (PSTs)

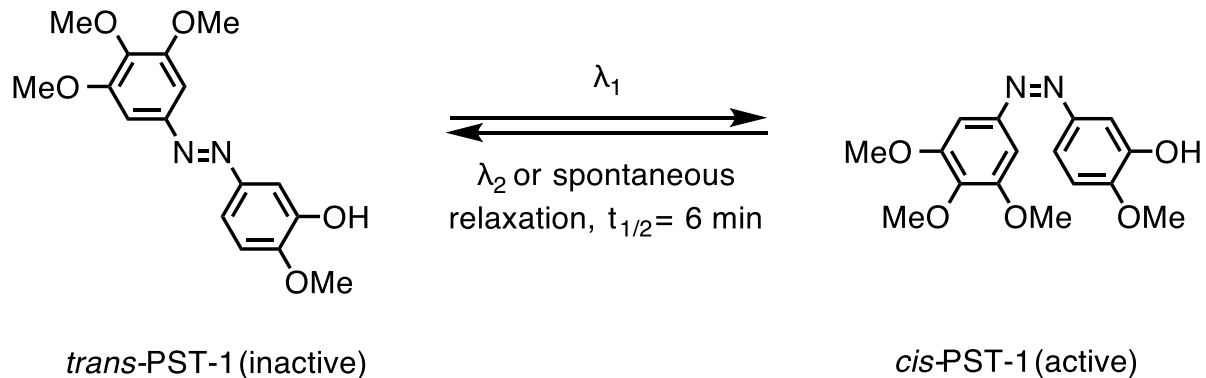
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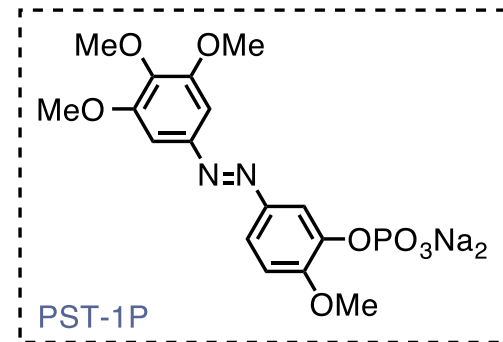
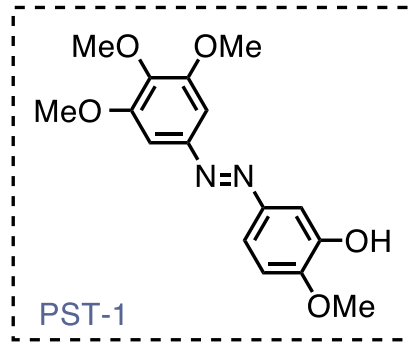
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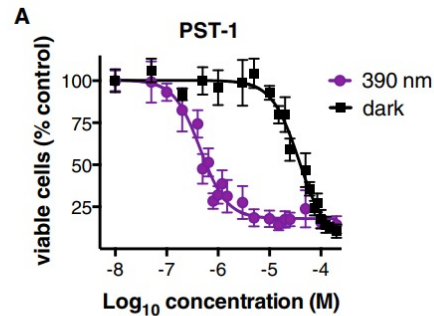
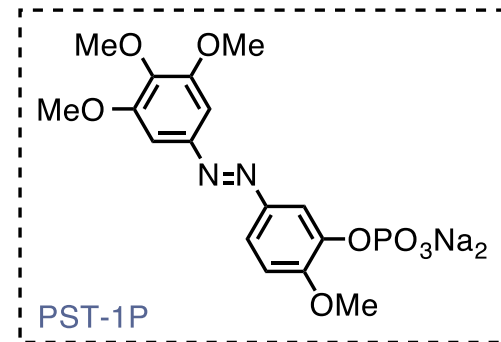
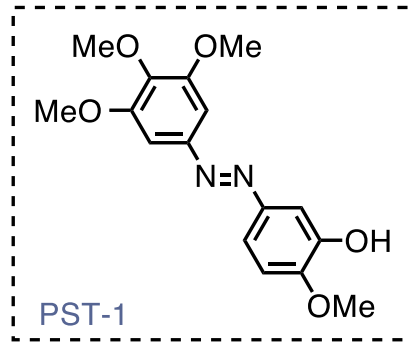
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# Photostatins (PSTs)



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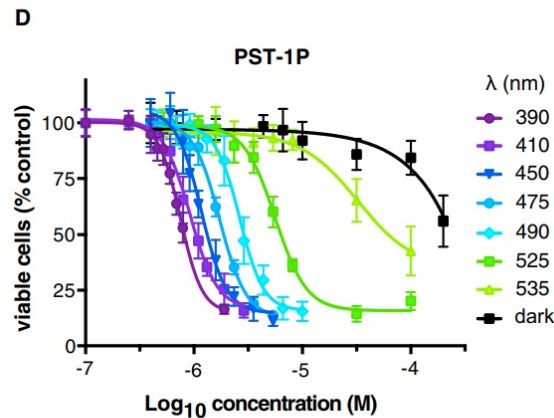
**B**

Compound	EC <sub>50</sub> dark	EC <sub>50</sub> 390 nm
PST-1	38 μM	0.5 μM
PST-2	25 μM	0.5 μM
PST-3	35 μM	2.1 μM
PST-4	43 μM	1.4 μM
PST-5	37 μM	1.0 μM
PST-1P	71 μM	0.7 μM
PST-2S	ND	5.4 μM
PST-1CL	ND	4.2 μM
CA4P	4 nM	-

□ *In cellulo* studies with MDA-MB-231 human breast cancer cells and HeLa (cervical cancer) cells

**C**

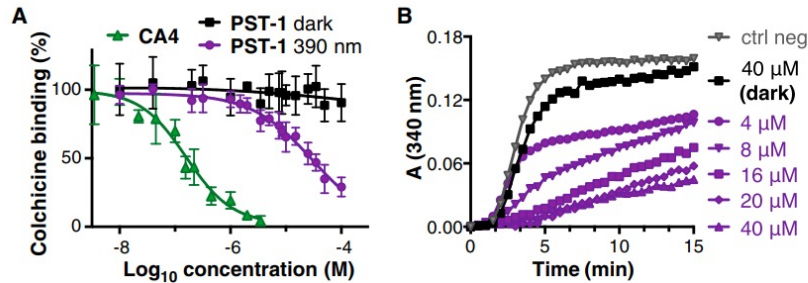
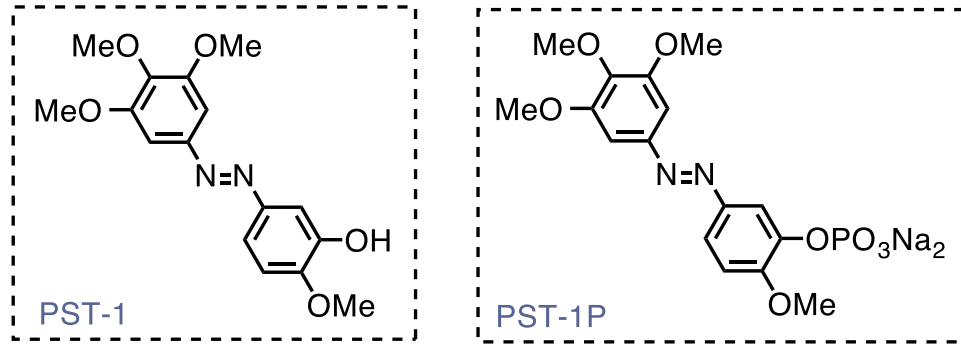
λ (nm)	EC <sub>50</sub> (μM)
370	0.97
380	0.70
390	0.75
400	0.91
410	0.93
435	0.98
450	1.2
465	1.5
475	1.7
490	2.6
505	2.6
515	3.0
525	5.6
535	32
dark	> 200



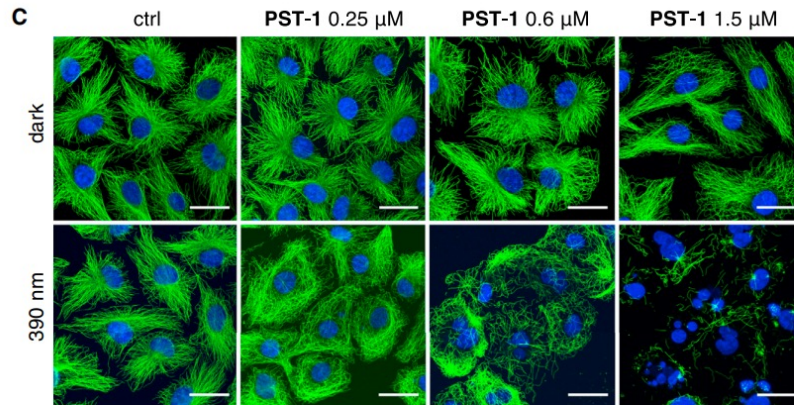
## Main takeaways:

- PSTs are more cytotoxic under 390 nm irradiation than in the dark
- PST-1P cytotoxicity can be modulated by applying light of different wavelengths

# Photostatins (PSTs)



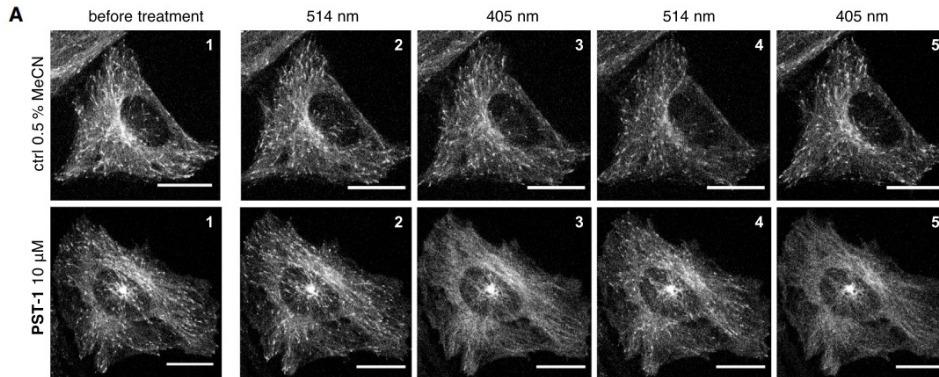
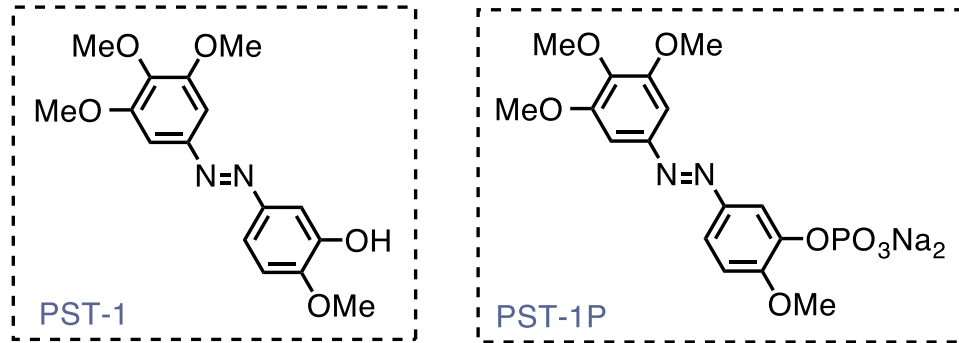
*In vitro and in cellulo studies validating of tubulin as the molecular target of cis-PSTs*



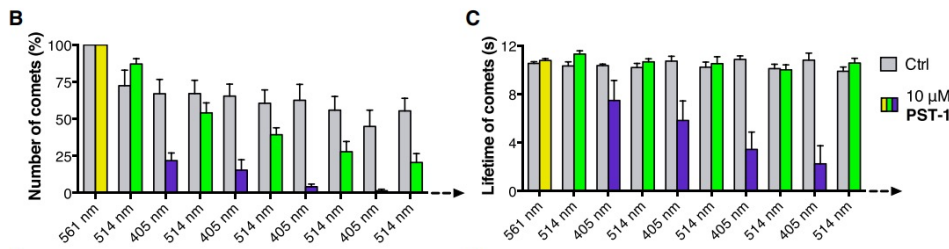
## Main takeaways:

- PST-1 exposed to 390 nm light competes with colchicine for tubulin binding
- PST-1 exposed to 390 nm light dose-dependently induces microtubule breakdown

# Photostatins (PSTs)



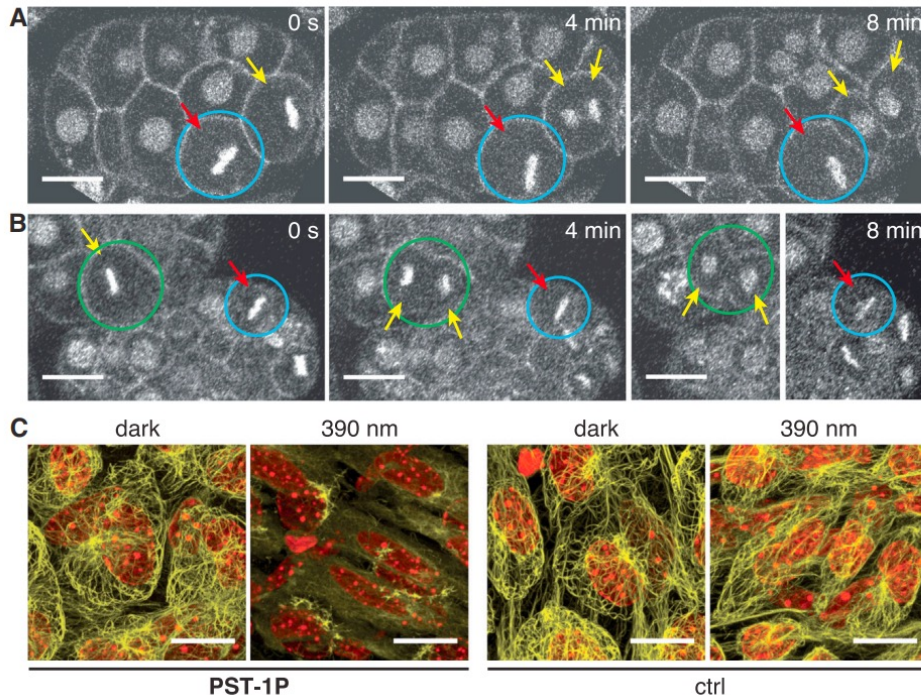
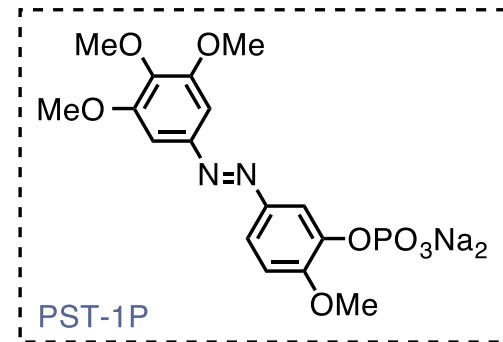
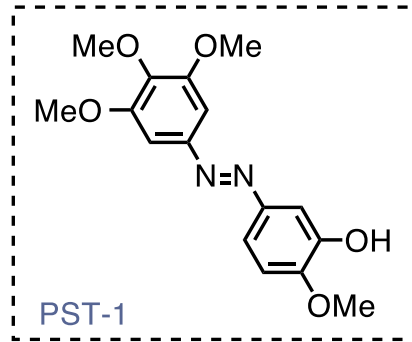
□ *In cellulo* studies imaging end-binding protein EB3, which clusters at the plus tips of growing MTs and dissociates in phases of MT shrinkage



Main takeaways:

- *Cis-trans* photoisomerization of PST-1 by alternating blue and green light causes MT to stop and start again with <1s response time with full reversibility

# Photostatins (PSTs)



□ *In vivo studies with C. elegans embryo and live mouse muscle tissue*

○ 405 nm light

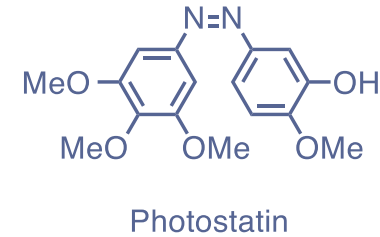
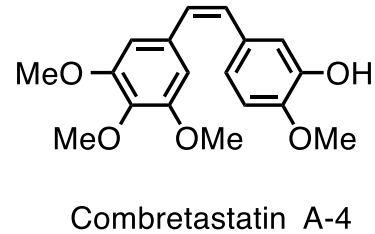
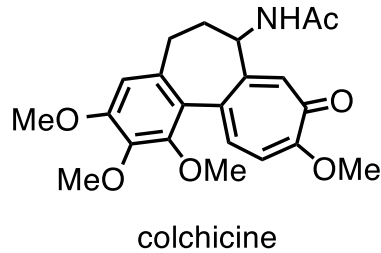
○ 405 nm + 514 nm light

Main takeaways:

- PSTscan be operated with high spatial control on the single-cell level

- PSTsexhibit reversible *in vivo* control over MT dyanmics

# Photostatins (PSTs)



## In summary:

- The *cis* isomer of PSTs are 250x more cytotoxic than the corresponding *trans* isomer
- PSTs can modulate mitosis with spatial precision on the cellular level
- PSTs can modulate microtubule activity on the time-scale of seconds
- PSTs can switch microtubule activity on and off by simply using blue & green light

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- Molecular Machines
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# Molecular Machines

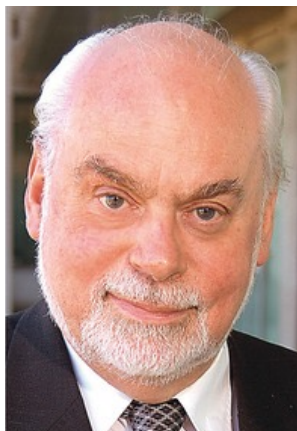
Molecular machines are molecules that convert chemical energy into mechanical forces and motion.



2016 Nobel Prize in Chemistry  
*"for the synthesis and design of molecular machines"*



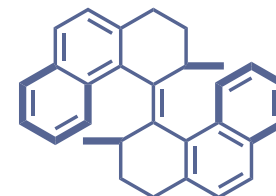
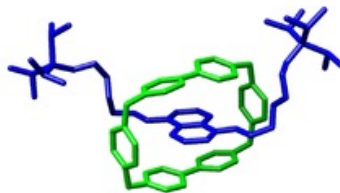
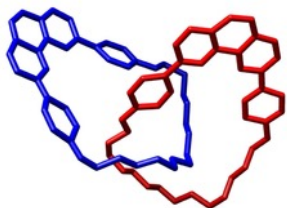
Jean-Pierre Sauvage



Sir J. Fraser Stoddart

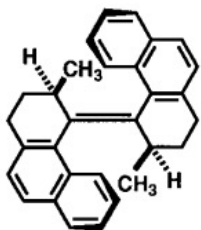


Bernard L. Feringa

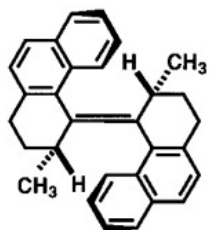


# First Generation Molecular Motor

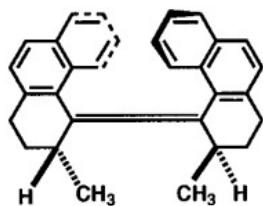
Biphenanthrylidene scaffold with axial chirality  
"helicity"



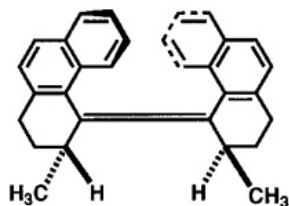
(3R,3'R)-(P,P)-(E)-3  
with two axial methyl groups  
C<sub>2</sub>-symmetry  
ΔE = 0.0 kcal/mol



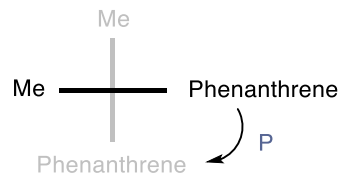
(3R,3'R)-(M,M)-(E)-3  
with two equatorial methyl groups  
C<sub>2</sub>-symmetry  
ΔE = +10.2 kcal/mol



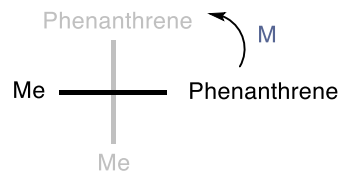
(3R,3'R)-(P,P)-(Z)-4  
with two axial methyl groups  
C<sub>2</sub>-symmetry  
ΔE = 0.0 kcal/mol



(3R,3'R)-(M,M)-(Z)-4  
ΔE = +11.1 kcal/mol



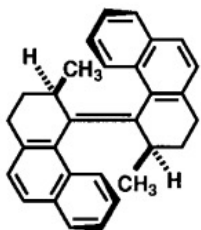
P-(E)



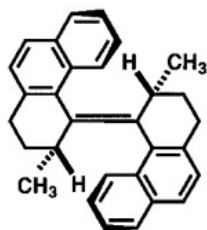
M-(E)

# First Generation Molecular Motor

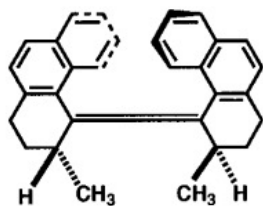
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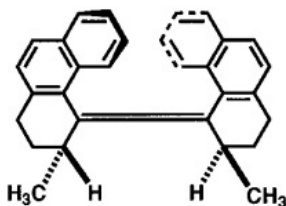
**(3R,3'R)-(P,P)-(E)-3**  
with two axial methyl groups  
C<sub>2</sub>-symmetry  
 $\Delta E = 0.0$  kcal/mol



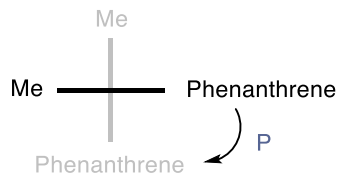
**(3R,3'R)-(M,M)-(E)-3**  
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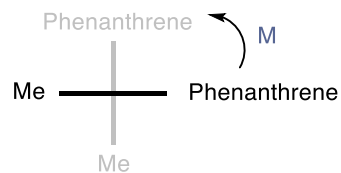
**(3R,3'R)-(P,P)-(Z)-4**  
with two axial methyl groups  
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 $\Delta E = 0.0$  kcal/mol



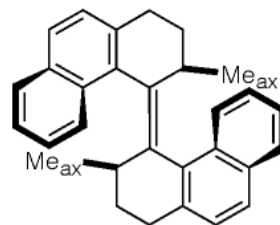
**(3R,3'R)-(M,M)-(Z)-4**  
 $\Delta E = +11.1$  kcal/mol



P-(E)

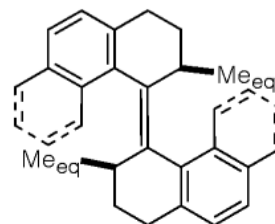


M-(E)

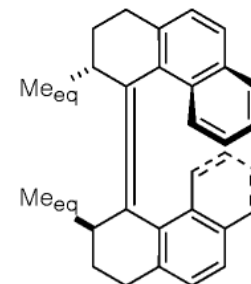
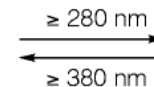


**(P,P)-trans-1**

60 °C

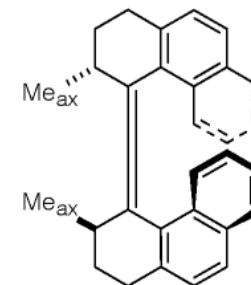


**(M,M)-trans-1**

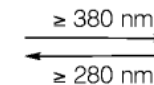


**(M,M)-cis-2**

20 °C

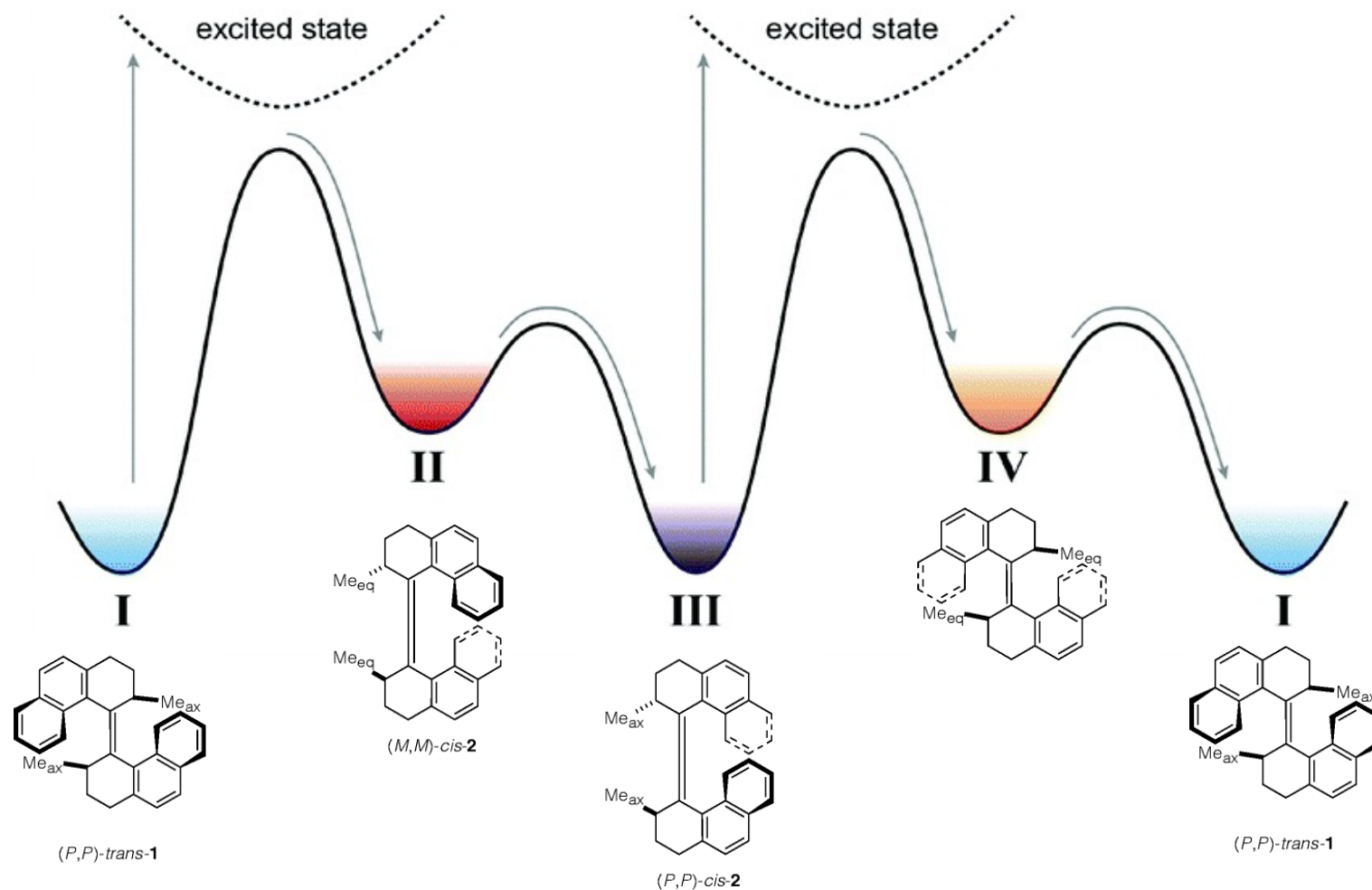


**(P,P)-cis-2**

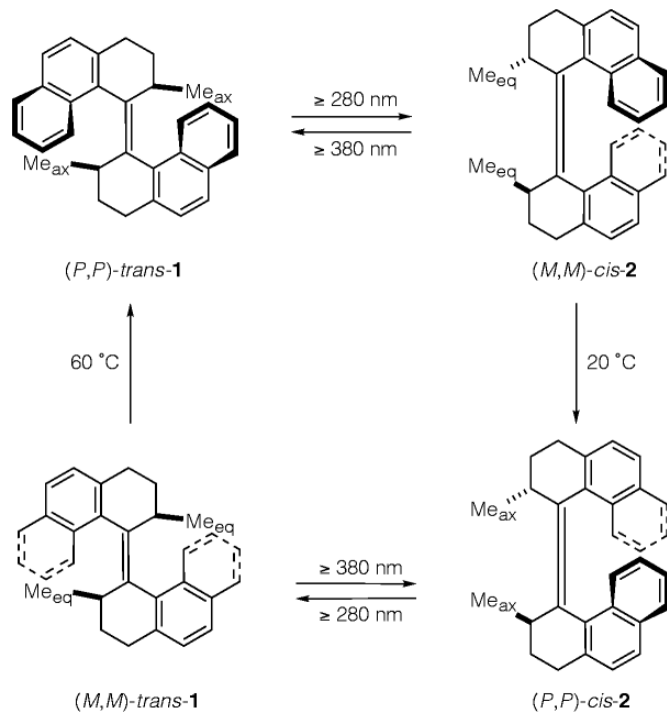


# First Generation Molecular Motor

The first molecular motor was developed on the basis of coupling photo- and thermodynamic equilibria.

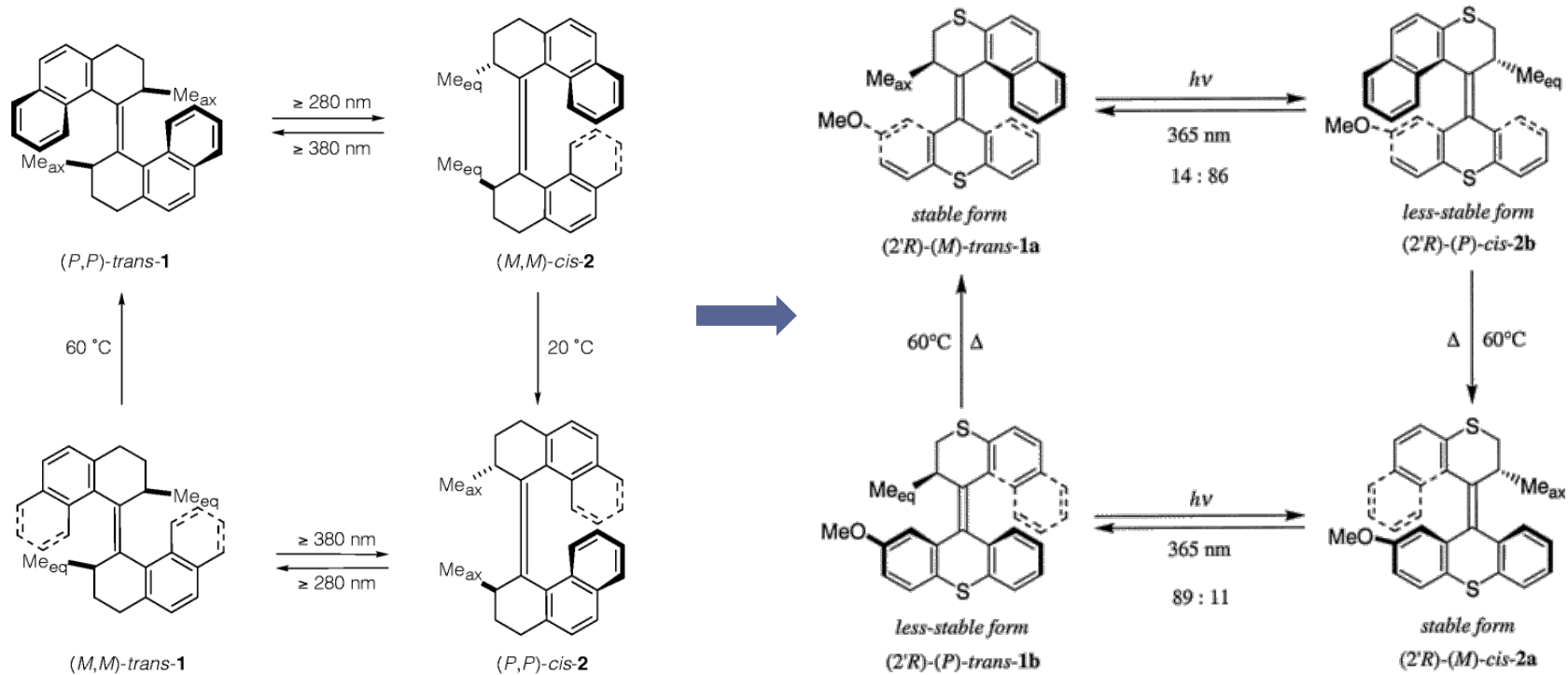


# First Generation Molecular Motor



Continuous, unidirectional,  $360^\circ$  rotation

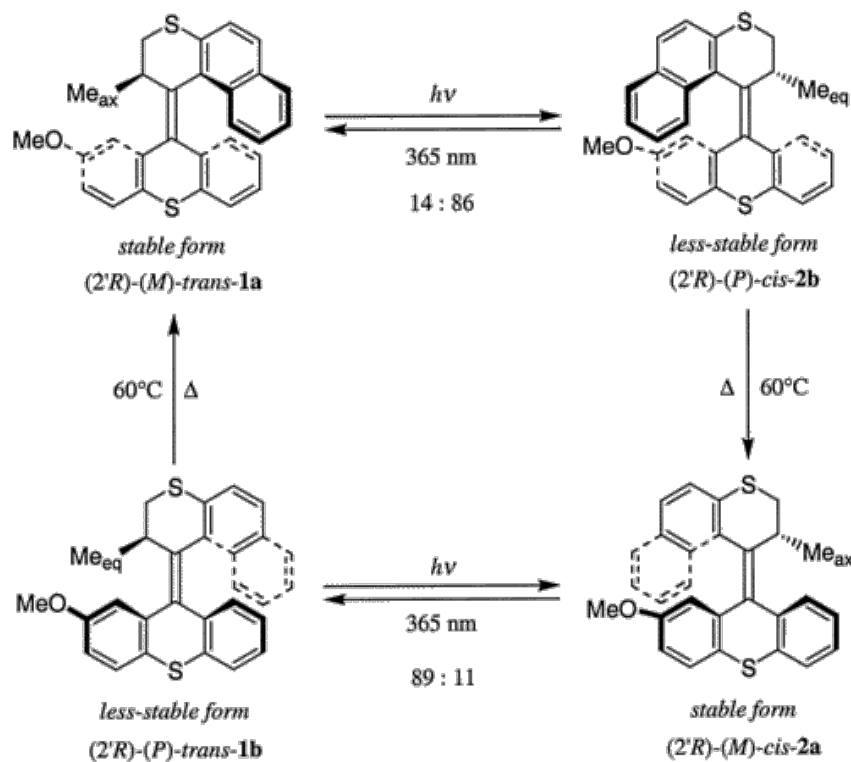
# Second Generation Molecular Motor



Continuous, unidirectional,  $360^\circ$  rotation

Asymmetric scaffold, more control elements

# Second Generation Molecular Motor



## Design Principles

- Stilbene scaffold
- Distinct upper and lower parts
- Introduction of stereogenic center enables control over direction of rotary motion
- Introduction of bridging atoms enables control over speed of rotary motion

# Outline

## I. Introduction

## II. Applications

- Photopharmacology
- Molecular Machines
- Solar Energy Storage

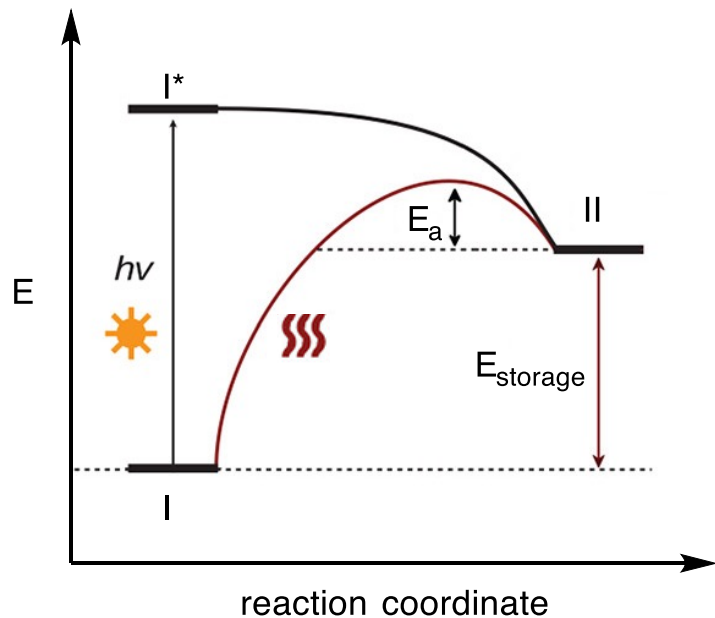
## III. Conclusion & Outlook



# Solar Energy Storage

Molecular solar thermal (MOST) systems are based on use of photoswitches for energy storage.

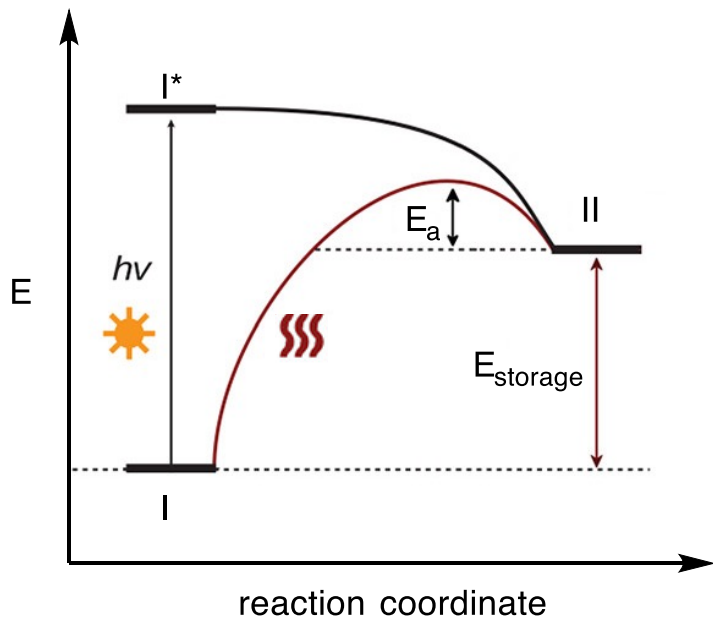
Molecular photoswitches employed in this sense are also known as solar thermal fuels (STF).



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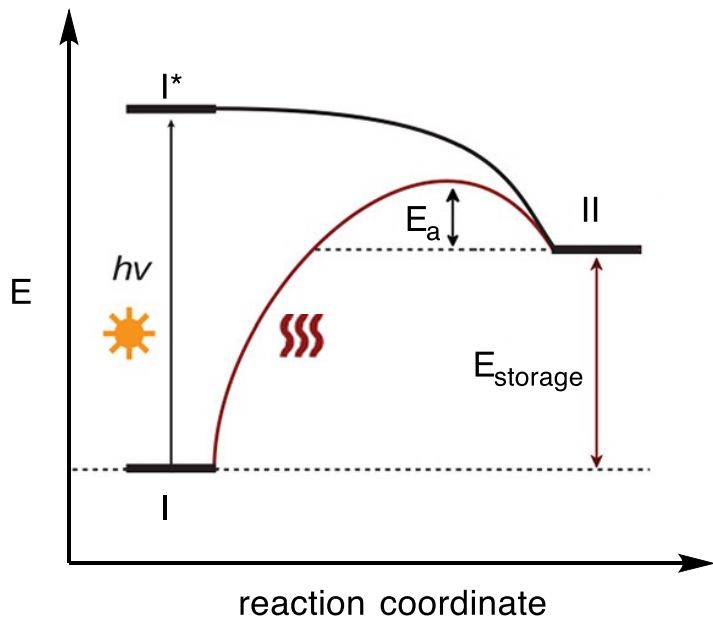
Ideal systems should have:

- an absorption onset of 300-600 nm
- high quantum yields of isomerization from I  $\rightarrow$  II
- low molecular weight (energy density >0.3 MJ/kg)
- a serviceable half life ( $t_{1/2}$ )

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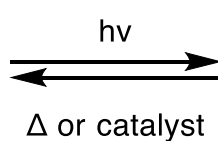


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norbornadiene  
(NBD)



quadricyclane  
(QC)

$$\lambda_{\max}(\text{NBD}) = 213 \text{ nm}$$

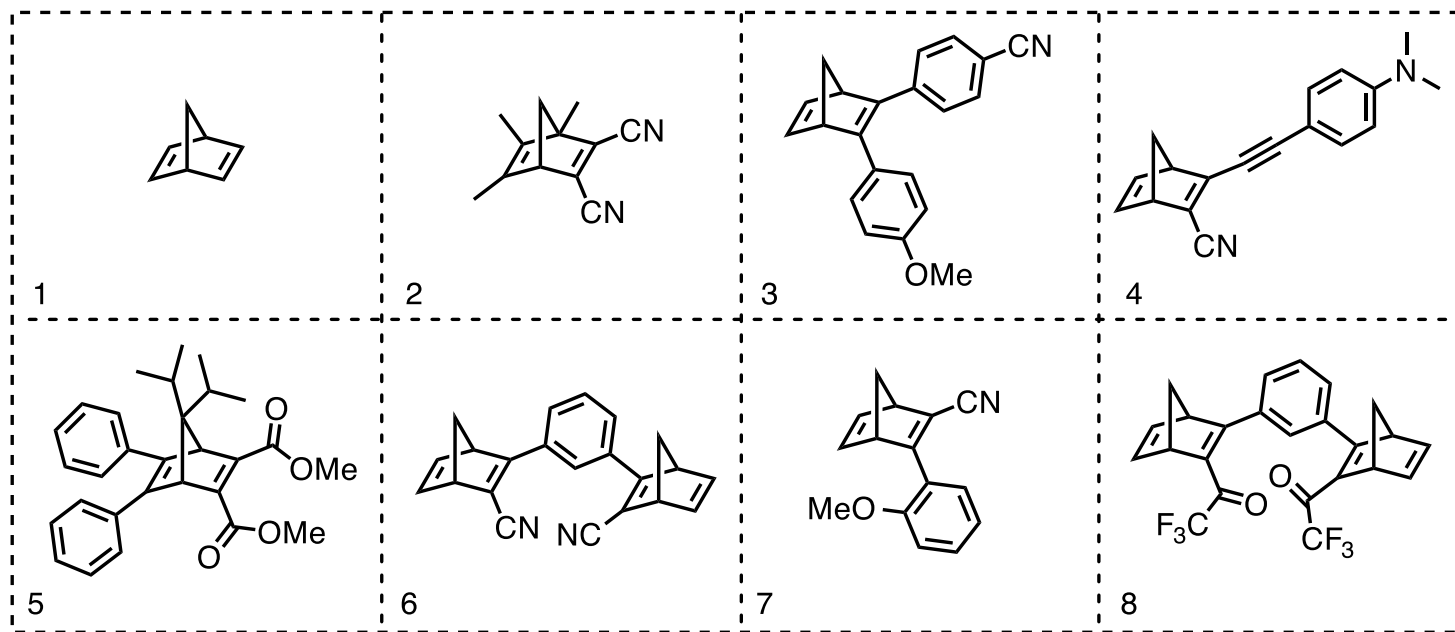
$$\lambda_{\max}(\text{QC}) = 236 \text{ nm}$$

$$\text{MW} = 92 \text{ g/mol}$$

$$\Delta G_{\text{isom}}(\text{QC} \rightarrow \text{NBD}) = 89 \text{ kcal/mol}$$

$$t_{1/2}(\text{QC}) = \sim 10^7 \text{ h (1140 y) @ RT}$$

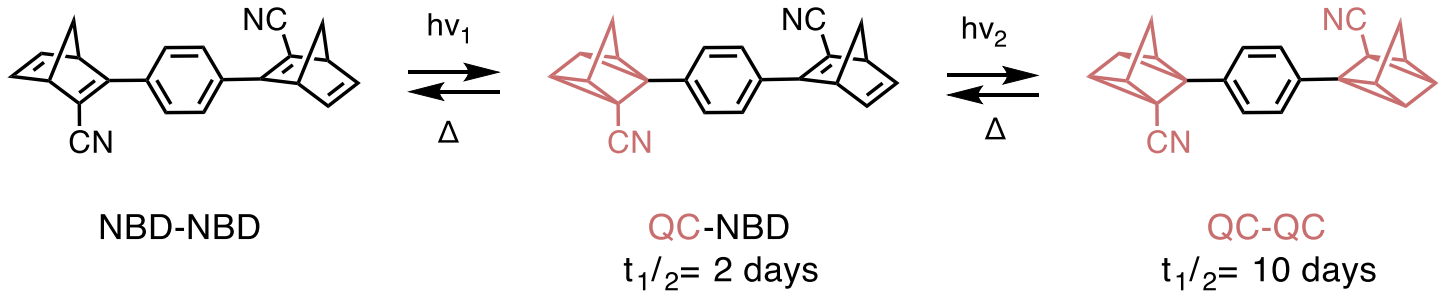
# NBD Modifications: Red Shifting and QC Stabilization



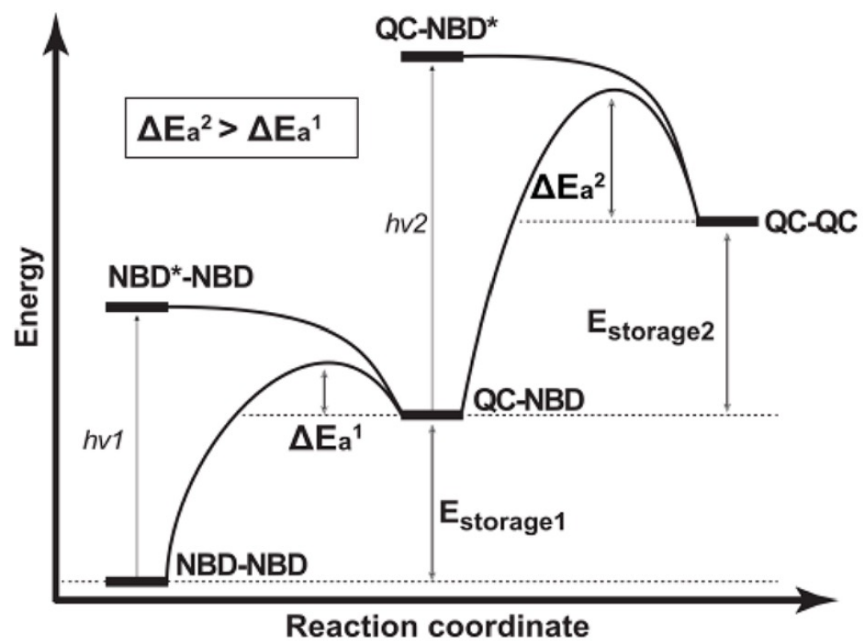
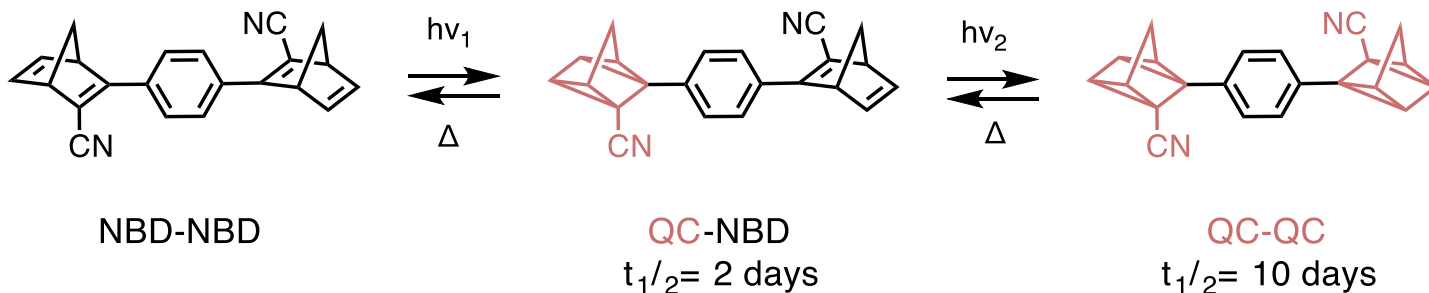
**Table 1. MOST Parameters of NBDs 1–8**

NBD	MW (g/mol)	$\lambda_{\text{onset}}^a$ (nm)	$\varphi$	$t_{1/2}^b$ (days)	$\Delta E_{\text{storage}}$ (kJ/mol)
<b>1</b> <sup>23,37</sup>	92	300	0.05		89
<b>2</b> <sup>24</sup>	184	360 <sup>c</sup>	0.96	124 <sup>d</sup>	88
<b>3</b> <sup>13</sup>	299	431	0.62	8.7	
<b>4</b> <sup>14</sup>	260	456	0.28	0.2	103
<b>5</b> <sup>11</sup>	445	414	0.88	2.3	46
<b>6</b> <sup>38</sup>	308	362	0.53	49	173
<b>7</b> <sup>2</sup>	223	368	0.70	2273	89
<b>8</b> <sup>4</sup>	450	466	0.77	0.7	216

# NBD Dimer

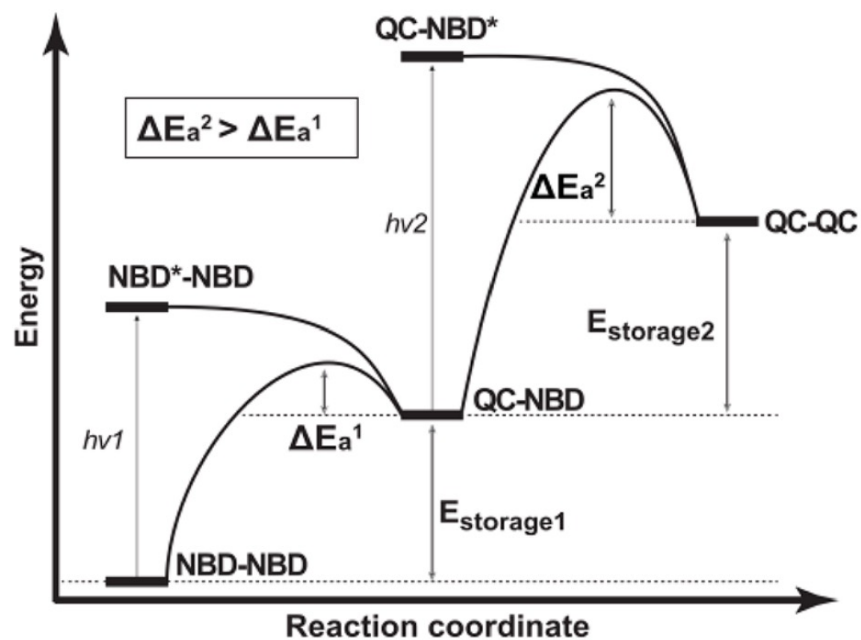
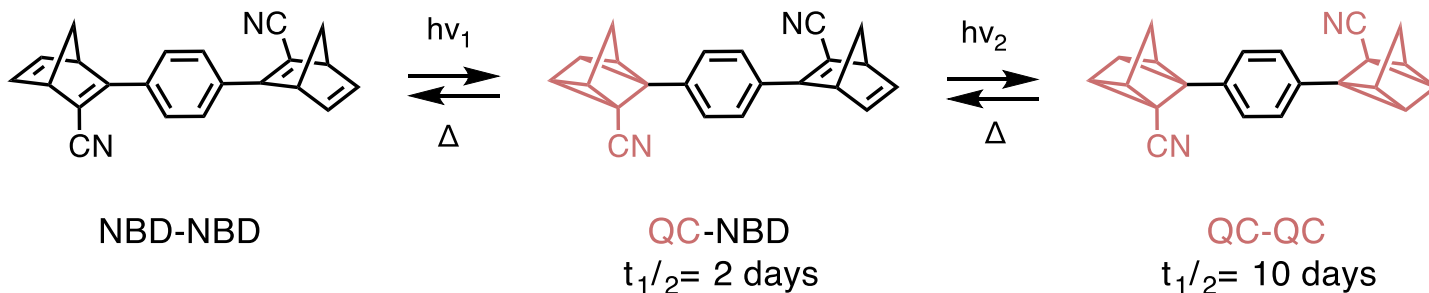


# NBD Dimer

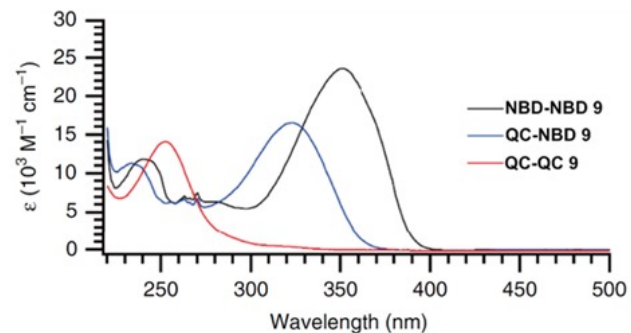


energy density > 0.3 MJ/kg  
 dimer & trimer systems showed energy densities up to 0.9 MJ/kg

# NBD Dimer

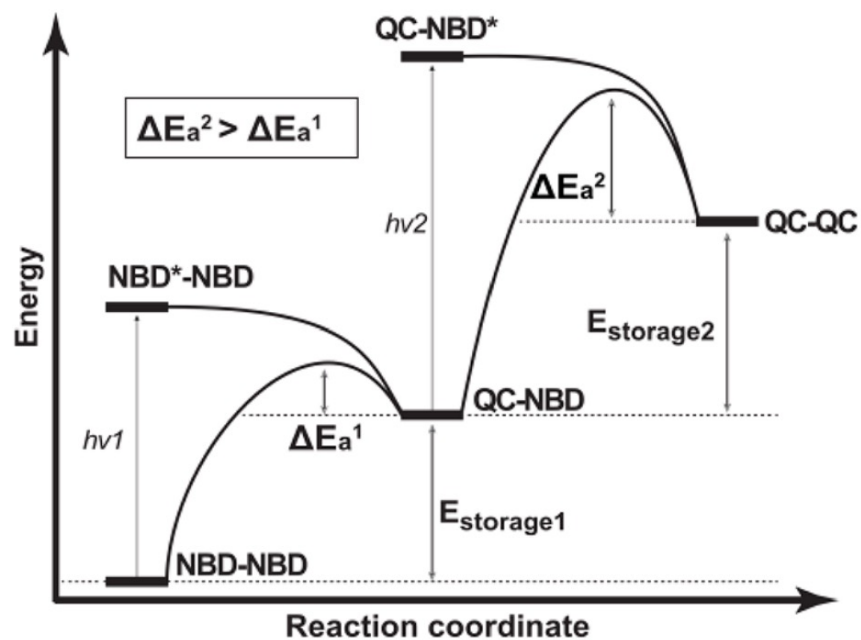
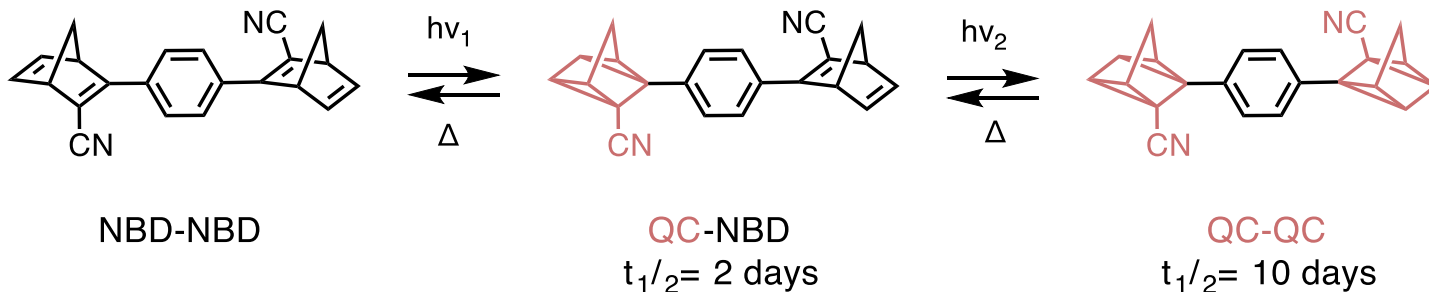


UV-Vis of dimer isomers



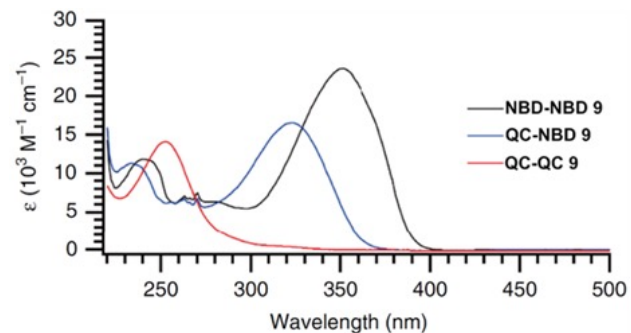
energy density  $> 0.3$  MJ/kg  
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# NBD Dimer

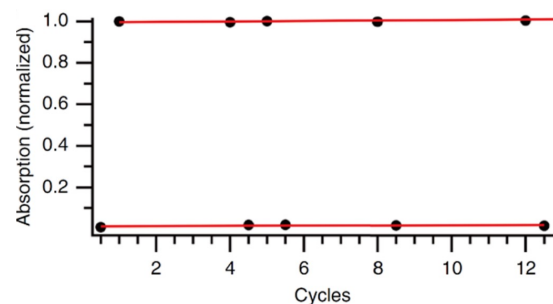


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UV-Vis of dimer isomers

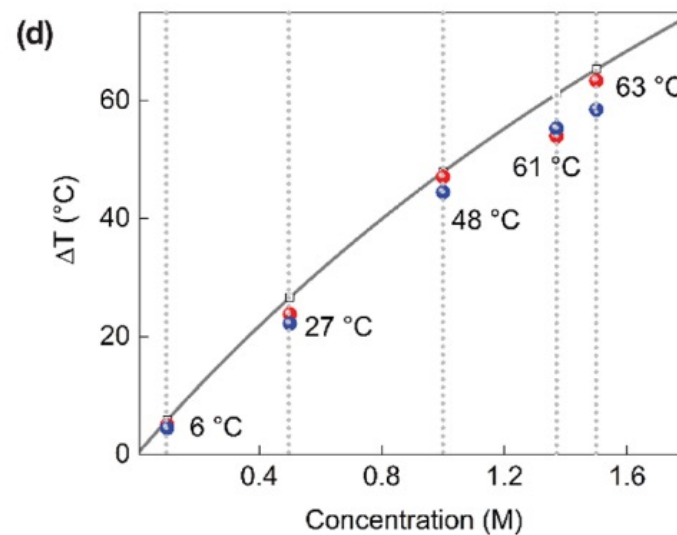
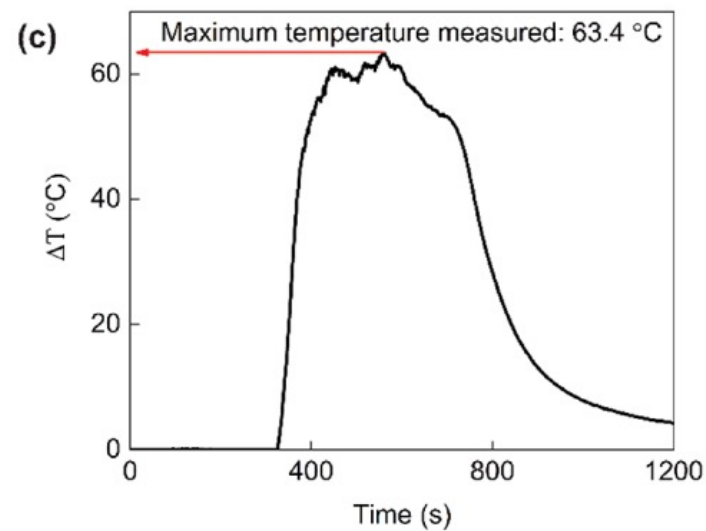
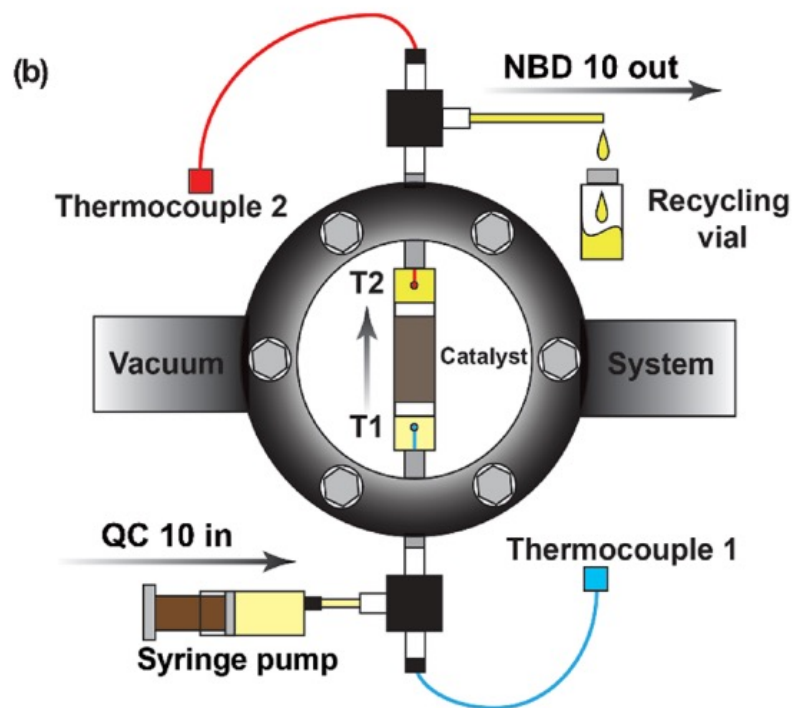
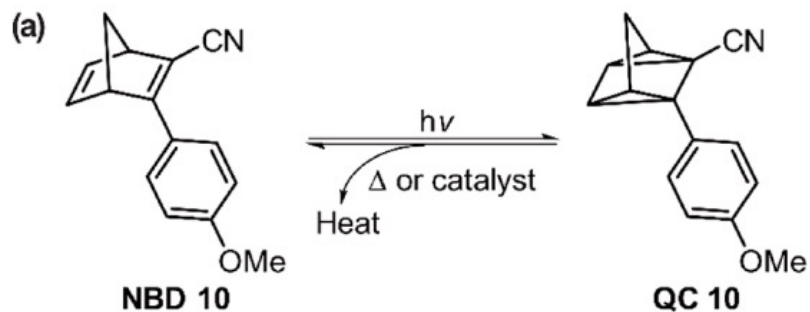


Degradation per cycle = 0.11%





# Designing Prototypical Storage Devices



# Outline

I. Introduction

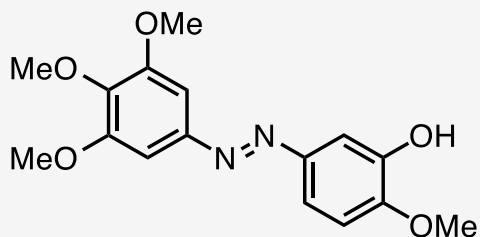
II. Applications

- Photopharmacology
- Molecular Machines
- Solar Energy Storage

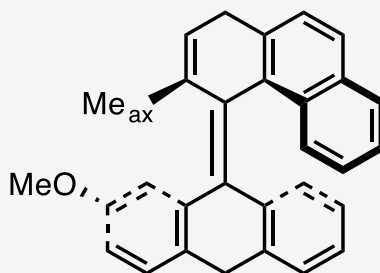
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# Conclusion & Outlook

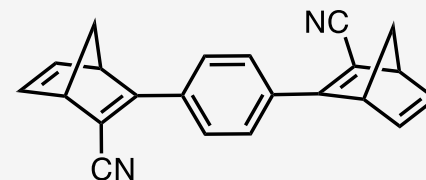
- Molecular photoswitches are a class of small molecules that can undergo a reversible change in their physical and chemical properties upon exposure to light.
- Photoswitches are highly versatile, as they can be triggered by different wavelengths of light, and can be designed to switch between various states, such as between an "on" and "off" state.
- They have a wide range of potential applications, including those in drug delivery, biomaterials, molecular machines, data and energy storage, sensing, and more !



Photopharmacology



Molecular motors



Solar energy storage

- The ideal photoswitch is the photoswitch that has been optimized for the task at hand.
- Future directions in this field looks towards visible light and NIR activation, improving photoefficiency, and systems including multiple photoswitches.

*Thank you for your attention !*